# Molecular Physics (Prof. Käs)

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### **Problem Set 13**

Due date: January 28, 2008

## Problem 48)

Estimate the maximum laser power of a cylindric ruby crystal with 4.0 cm length and 0.6 cm diameter in a pulse of 120 ns duration ( $\lambda = 694.3$  nm). The ruby consists of 0.050 Cr<sup>3+</sup> mass percent within the Al<sub>2</sub>O<sub>3</sub> matrix with an overall density of 3.97 g/cm<sup>3</sup>. Assume that the pumping radiation is sufficient to pump all chromium ions out of the ground state at a rate faster than they decay back to the ground state.

(4 points)

## Problem 49)

Evaluate the results of a rotation-vibration spectrum with two spectral branches (infrared) given in  $cm^{-1}$ :

J	0	1	2	3	4	5	6
$^{1}\text{H}^{35}\text{Cl}$	2906.25	2925.92	2944.99	2963.35	2981.05	2998.05	3014.50
(R branch)							
<sup>1</sup> H <sup>35</sup> Cl		2865.14	2843.63	2821.59	2799.00	2775.77	2752.01
(P branch)							

For the energy levels (in units of the wavenumber), the following relation holds:

$$S(v,J) = \left(v + \frac{1}{2}\right)\widetilde{v} + B_{v}J(J+1)$$

The lines of the rotation-vibration spectrum appear at  $\Delta S_J^{P,R,Q} = S(v+1,J') - S(v,J'')$ . For the P branch holds J=J''-1, for the R branch holds J=J''+1, and for the Q branch hold J=J''.

Derive an expression for  $\Delta S_{J-1}^{R} - \Delta S_{J+1}^{P}$  in the general case and make then use the values given in the table in order to calculate a mean for value for  $B_{\nu}$ . Use the reduced mass  $\mu(^{1}H^{35}Cl)=1,6266\cdot10^{-27}$  kg.

(7 points)

## Problem 50)

To describe the intramolecular energy of a diatomic molecule the Morse potential is used in the form  $V(x) = D_e \cdot (1 - e^{-a \cdot x})^2$ , with  $x = r - r_e$  and  $r_e$ , the equilibrium separation of the nuclei.  $D_e$  is the (expectence of the separation of the nuclei) discovery of the separation of the nuclei.

is the (spectroscopic) dissociation energy and the constant *a* is defined as  $a = \sqrt{\frac{\mu_{red}}{2D_e}} \omega$ .

The values for the H<sub>2</sub> molecule are:  $r_e = 74.1$  pm,  $D_e = 7.61 \times 10^{-19}$  J and a = 0.0193 (pm)<sup>-1</sup>.

a) Display V(x) for H<sub>2</sub> in the range  $-50 \text{ pm} < r - r_e < 50 \text{ pm}$  graphically.

b) Approximate the Morse potential close to  $r_e$  by a parabola and use the latter to estimate the relevant force constant k, the classical vibration frequency (in cm<sup>-1</sup>) and the zero-level energy of the quantized oscillator.

(7 points)