

Exercises for Experimental Physics 4 – IPSP

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Exercise Sheet 7 (Summer Term 2013)

Date of Issue to Students: May 28th 2013

Date of Submission: June 4th 2013

Submission Place: Marked mailbox next to room 302 (Linnestr. 5)

Submission Time: 11:00 a.m. at the submission day noted above

Please note: Write your name and matriculation number on EACH sheet of paper. Only submit the calculations and results for exercise 1-3, exercise 4 will be discussed during the instruction classes.

Exercises:

1. (a) Given is the time-dependent Schrödinger Equation (SE)

$$i\hbar \frac{\partial}{\partial t} \Psi(\vec{x}, t) = \left(\frac{-\hbar^2}{2m} \nabla^2 + V(\vec{x}, t) \right) \Psi(\vec{x}, t).$$

Let V be time-independent. Use the separation-ansatz $\Psi(\vec{x}, t) = \Phi(\vec{x})T(t)$ to obtain the time-independent (stationary) SE and calculate $T(t)$. *Hint:* $f(\vec{x}) = g(t) = \text{const}$. Look up the solution for a free particle in order to determine the physical meaning of the freely chosen constant. (4 Points)

(b) Let $\Phi_1(x)$ and $\Phi_2(x)$ be solutions of the stationary SE. Show that $\Phi_3(x) = c_1\Phi_1(x) + c_2\Phi_2(x)$ is also a solution of the stationary SE. (4 Points)

2. The stationary wavefunction of the ground state of the hydrogen atom is $\Psi(r) = \frac{1}{\sqrt{\pi a_0^3}} e^{-r/a_0}$ with the Bohr radius $a_0 = 4\pi\epsilon_0\hbar^2/me^2$. Show that this wavefunction is indeed a solution of the stationary SE of the hydrogen atom. (a) Write down the SE for the hydrogen atom. Use spherical coordinates and the *Laplace operator in spherical coordinates*. (2 Points) (b) Plug in the wave function and energy of the ground state of hydrogen. (3 Points)

3. The probability distribution of a particle in a one-dimensional box (dimension: $0 < x < d$) is given by $P_{\text{classical}}(x) = 1/d$ and $P_{n,\text{quant}}(x) = |\Psi(x)|^2 = \frac{2}{d} \sin^2\left(\frac{n\pi x}{d}\right)$ with $n = 1, 2, 3, \dots$ for a classical and a quantum-mechanical particle, respectively.

(a) Calculate the expectation values $\langle x \rangle$ and $\langle x^2 \rangle$ for the classical and the quantum-mechanical particle in a box. (5 Points)

(b) What happens in the limit case of increasing n ? (2 Points)

Hint:

$$\int \sin^2(x) dx = \frac{x}{2} - \frac{1}{4} \sin(2x),$$

$$\int x \sin^2(x) dx = \frac{x^2}{4} - \frac{x}{4} \sin(2x) - \frac{1}{8} \cos(2x),$$

$$\int x^2 \sin^2(x) dx = \frac{x^3}{6} - \left(\frac{x^2}{4} - \frac{1}{8} \right) \sin(2x) - \frac{x}{4} \cos(2x)$$

4. An excited state of an electron decays exponentially with a certain life time τ by emitting a photon: $P(t) = e^{-|t|/\tau}$. This uncertainty in time leads to an uncertainty of the energy of the emitted photon.

(a) Calculate the shape of the emission spectrum of the photon by using Fourier transformation $\tilde{P}(\omega) = FT[P(t)](\omega)$

(b) Calculate the half width frequency $\omega_{1/2}$ where the intensity of the emission spectrum drops to $\frac{1}{2}\tilde{P}_{\text{MAX}}$