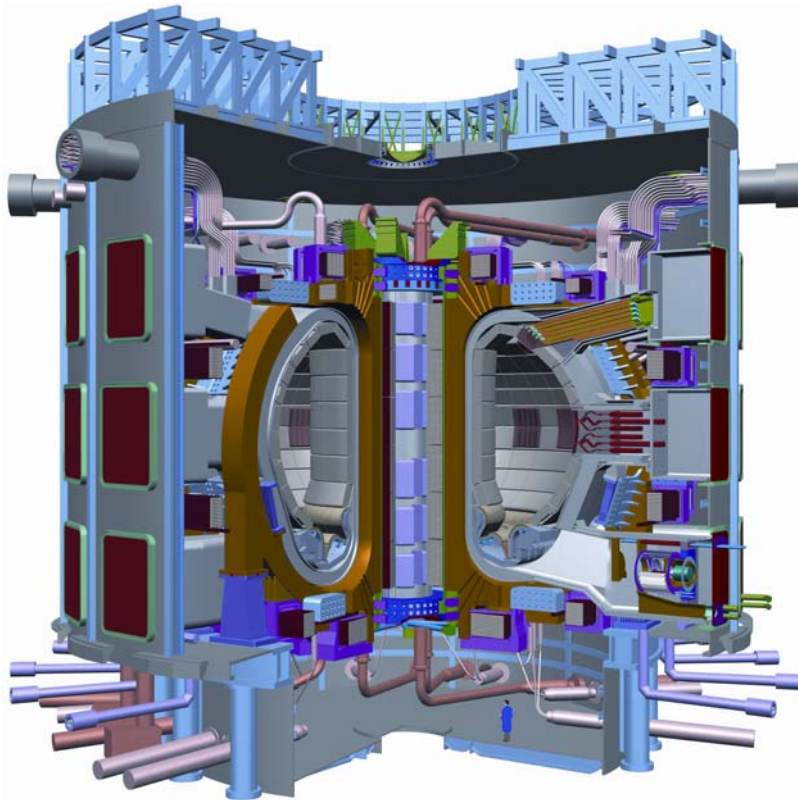
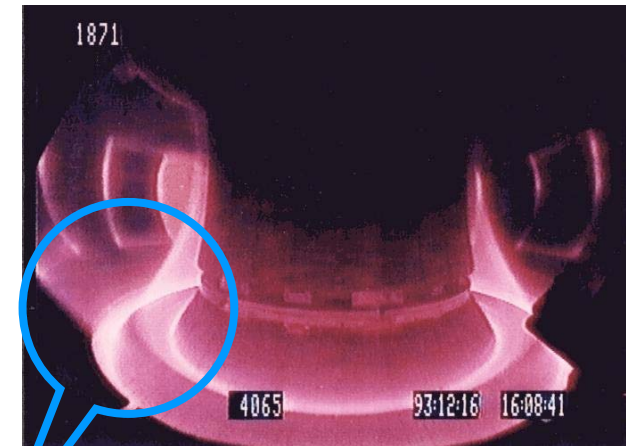


ITER: burning D/T plasma magnetically confined

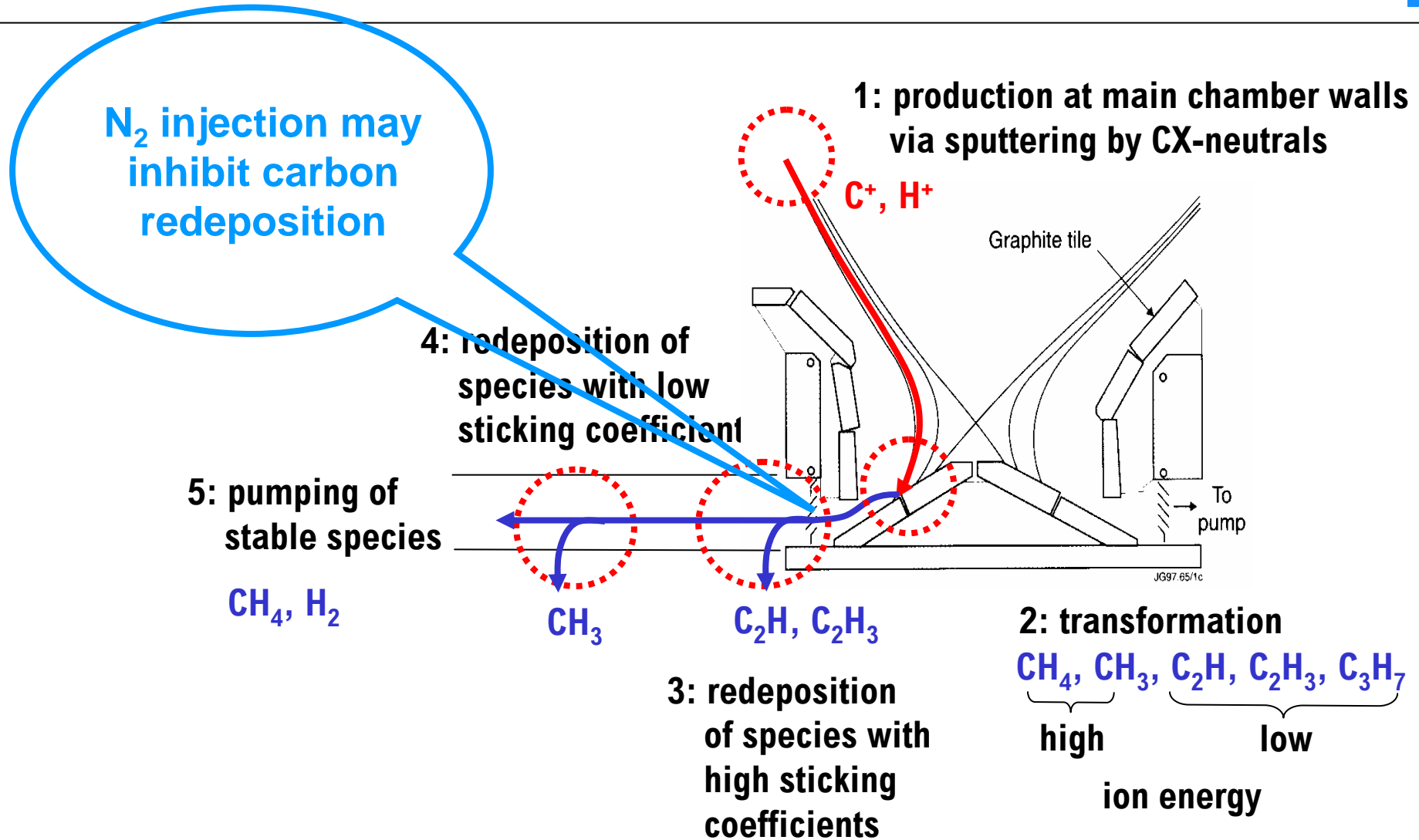


ASDEX Upgrade



strong plasma wall interaction
between
hot hydrogen plasma and wall
(wanted for various reasons)

carbonaceous walls: challenge of tritium redeposition



A. von Keudell et al. Nucl. Fusion 39, 10 (1999)

Stickstoffbeimischung in Kohlenwasserstoffplasmen - eine Lösung für die Redepositionsproblematik in Fusionsexperimenten?

NITROGEN PUFFING INTO HYDROCARBON PLASMAS –
A SOLUTION FOR REDEPOSITION IN FUSION EXPERIMENTS?

Thomas Schwarz-Selinger,

C. Hopf, M. Schlüter, W. Jacob

XIV. Erfahrungsaustausch
Oberflächentechnologie mit Plasma- und Ionenstrahlprozessen
Mühlleithen 13. März 2007

Outline:

- scavenger effect: the idea behind
 - a short reminder on a-C:H film growth
 - species in N_2 / CH_4 and N_2 / H_2 plasmas
 - film formation and erosion in N_2 / CH_4 and N_2 / H_2 plasmas
 - surface loss probabilities in N_2 / CH_4 plasmas
 - conclusion

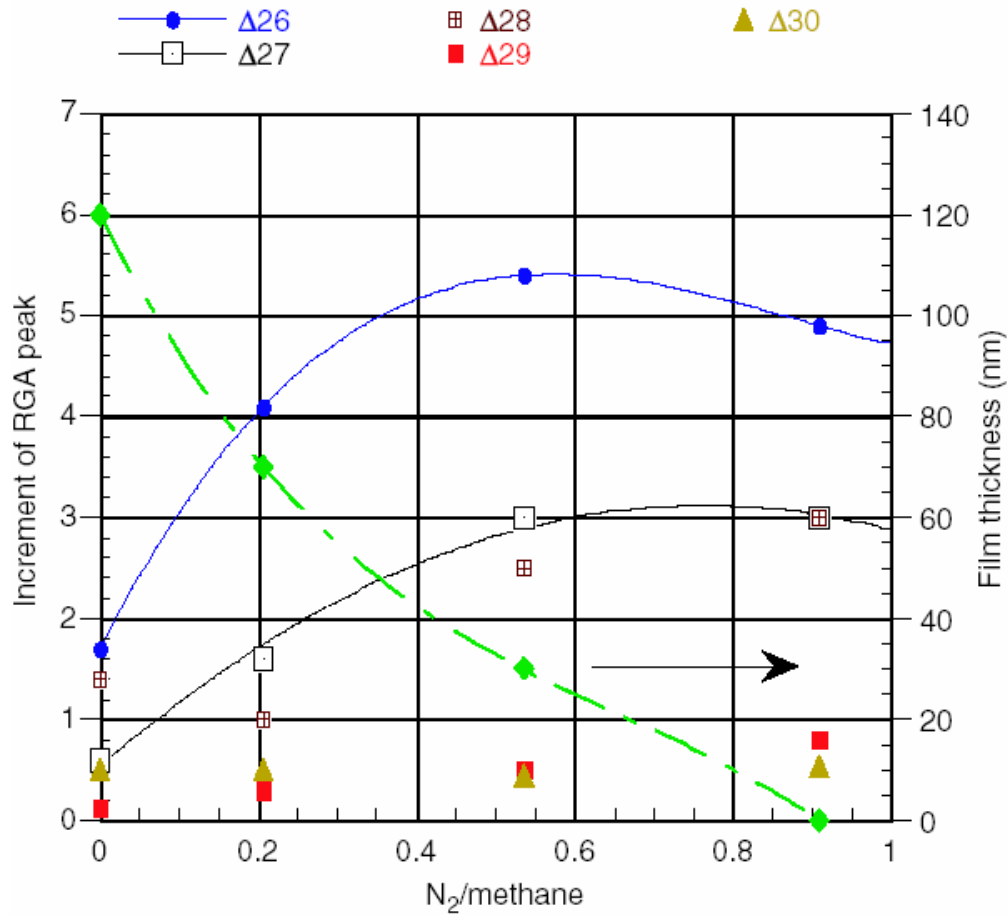
scavenger effect: the idea behind

F.L. Tabarés:

**use of scavenger techniques for the in-situ, shot-to-shot control
of the deposition of T/C layers in fusion devices**

(F.L. Tabarés et al., Plasma Phys. Control. Fusion 44, L37-42 (2002))

“**C₂H₂** and possibly **HCN** are the main stable gaseous compounds in N₂/CH₄ containing H₂ DC plasmas. Their relatively high melting points could allow trapping at liquid N₂ temperatures“



dc discharge
in hydrogen

F.L. Tabarés et al., Plasma Phys. Control. Fusion 44, L37-42 (2002)

the scavenger technique is based on

gas phase reactions of nitrogen (N , N_2 ; N_x^+) with the growth precursors (CH_3) forming

nonreactive species that can be pumped away or condensed on cold surfaces

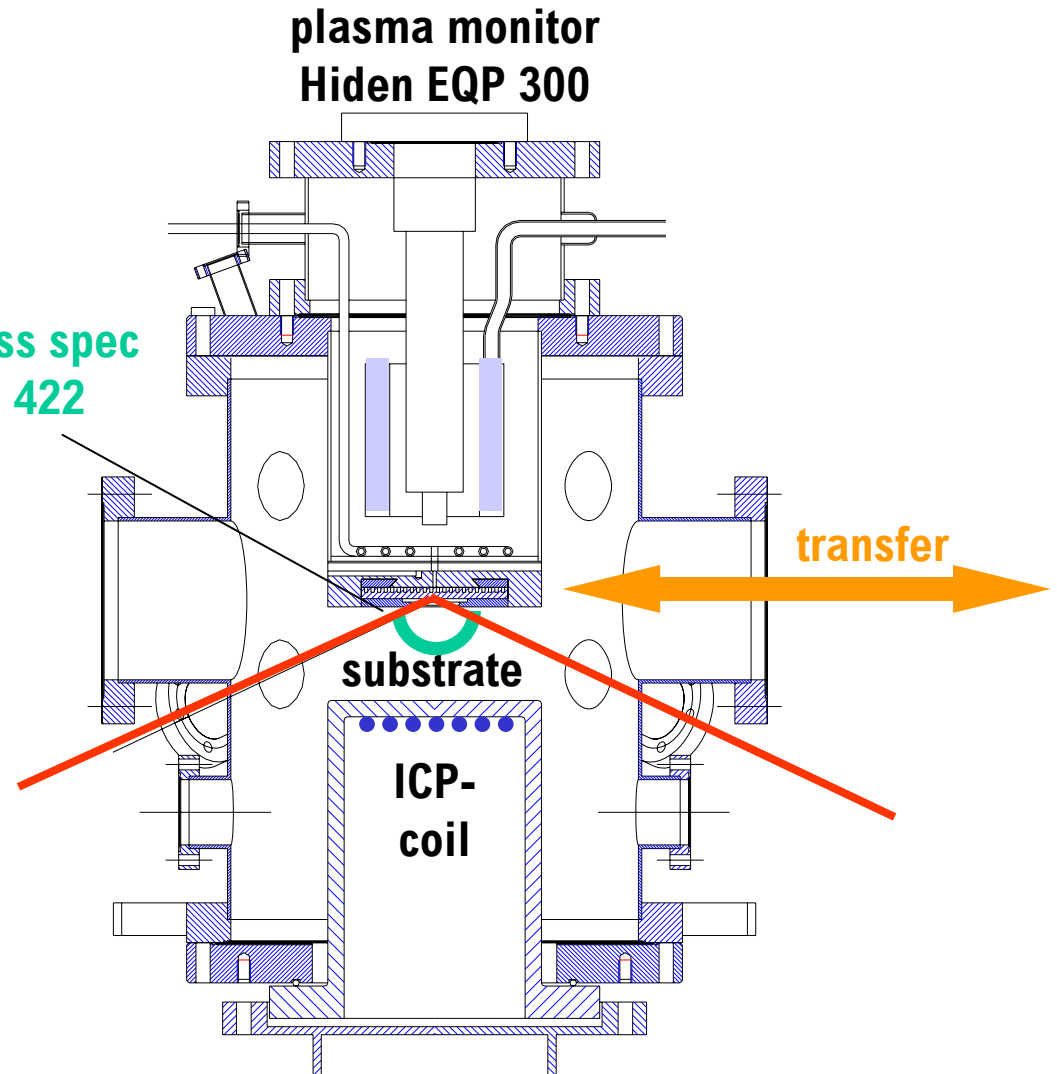
inductively coupled plasma source

pressure range: 1 - 50 Pa

rf power: < 600 Watt

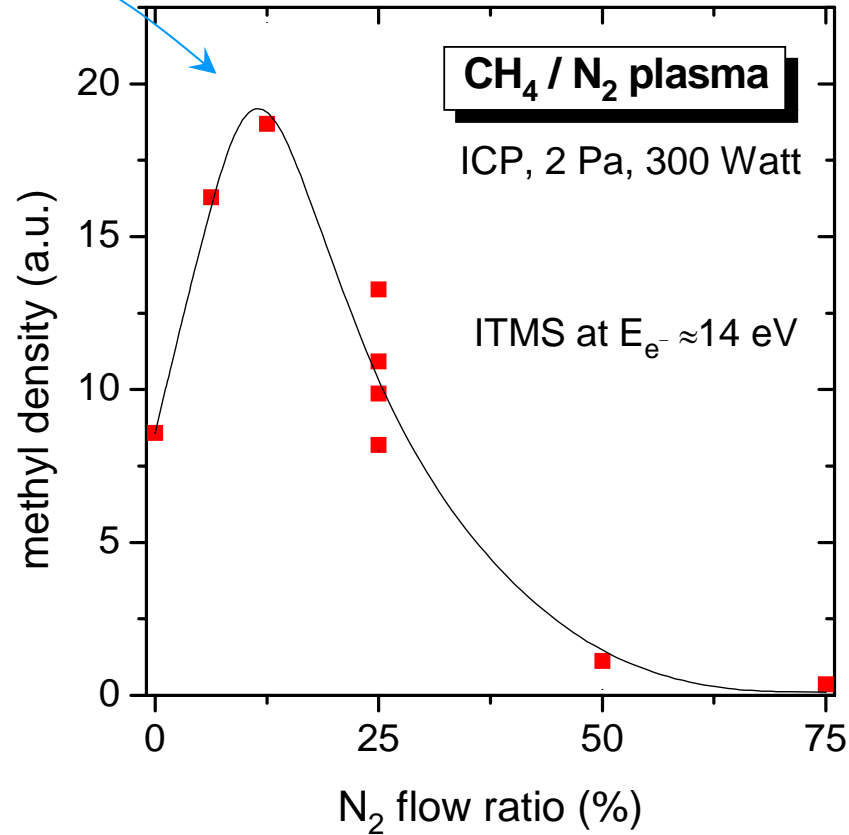
cross beam mass spec
Balzers QMG 422

ellipsometry,
FTIR



- Ionization Threshold Mass Spectrometry (ITMS):

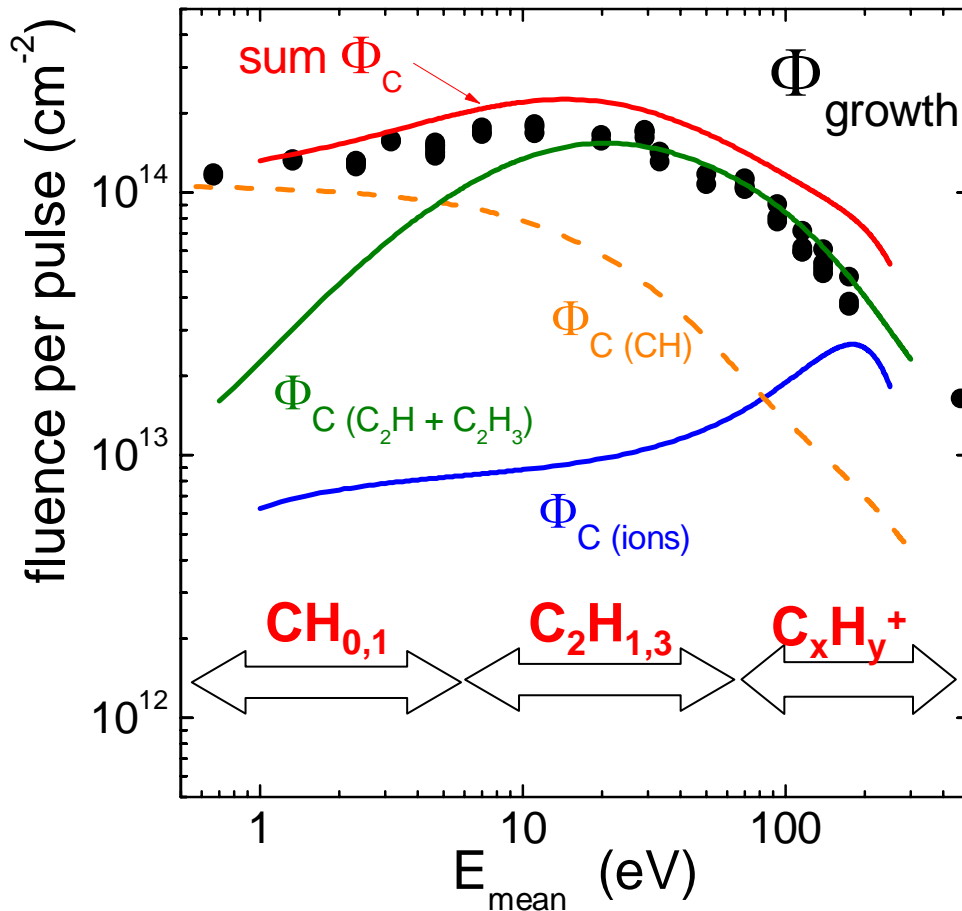
N₂ addition increases
CH₃ density!!!



fraction of NH⁺ negligible

a short reminder on a-C:H film growth

comparison of reactive flux (flux * sticking) with growth rate:

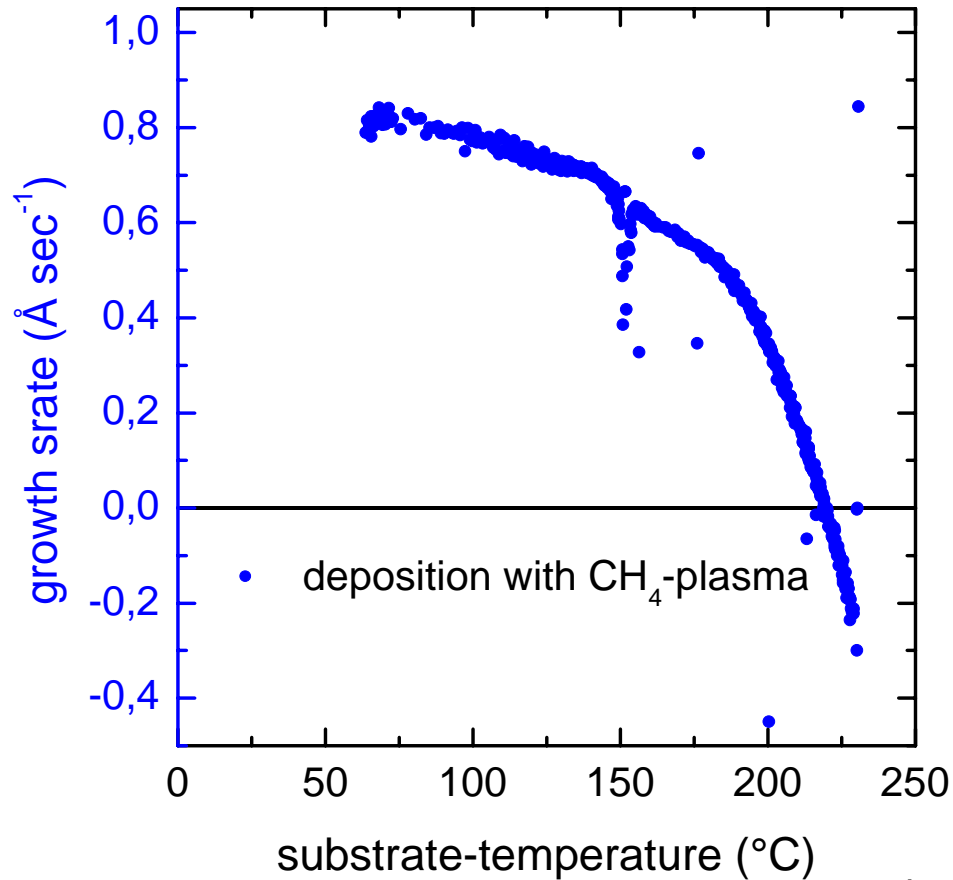


$$E_{\text{mean}} = \frac{P_{\text{mean}} \cdot \tau_{\text{residence}}}{N}$$

M. Bauer PhD 2004 (IPP report 9/135)

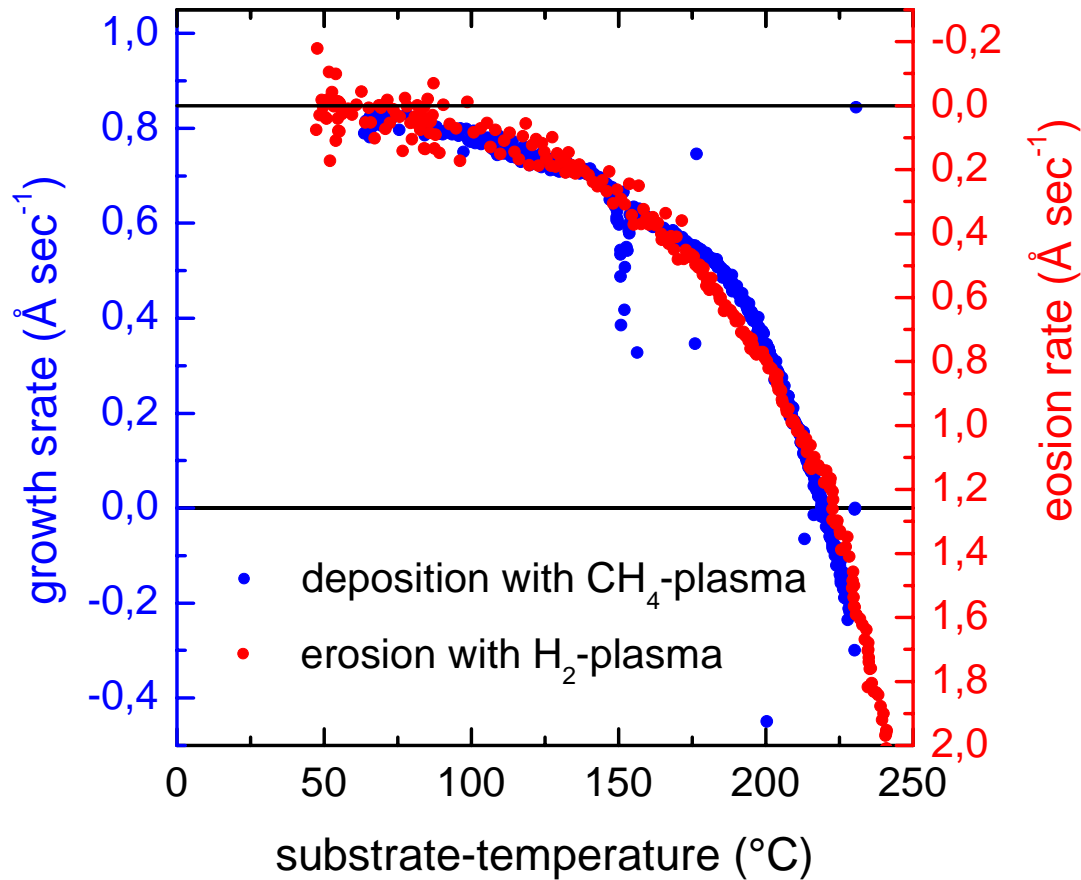
M. Bauer et al. Plasma Sources Sci. Technol. 14 (2005)

M. Bauer et al. J. Appl. Phys. 98, 2005



⇒ A. von Keudell et al. JAP 79 (2)1092 (1997).

growth rate = balance between deposition and erosion



⇒ A. von Keudell et al. JAP 79 (2)1092 (1997).

independent control of
substrate temperature and particle energy

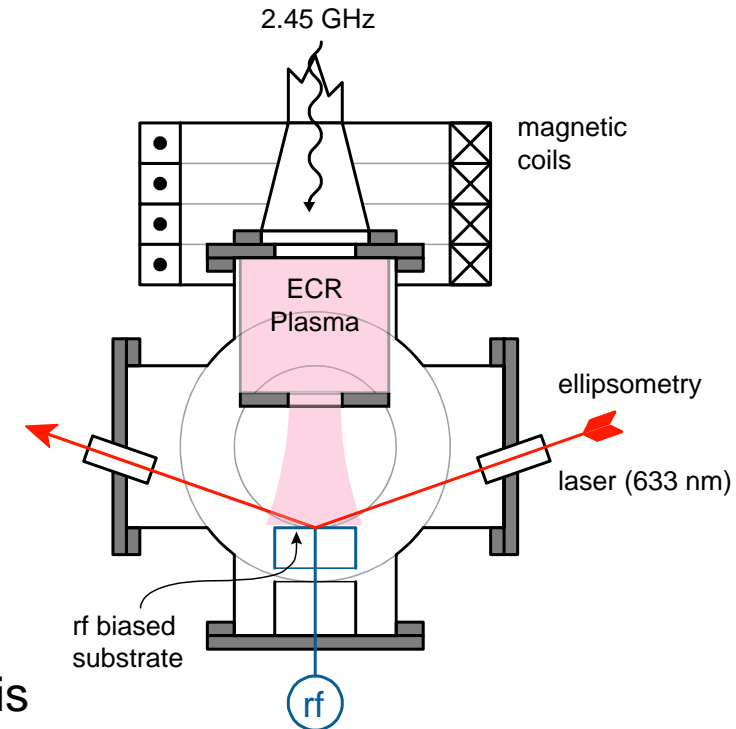
pressure range 10^{-2} - 1 Pa

ECR power: < 300 Watt

ellipsometry to detect erosion and deposition

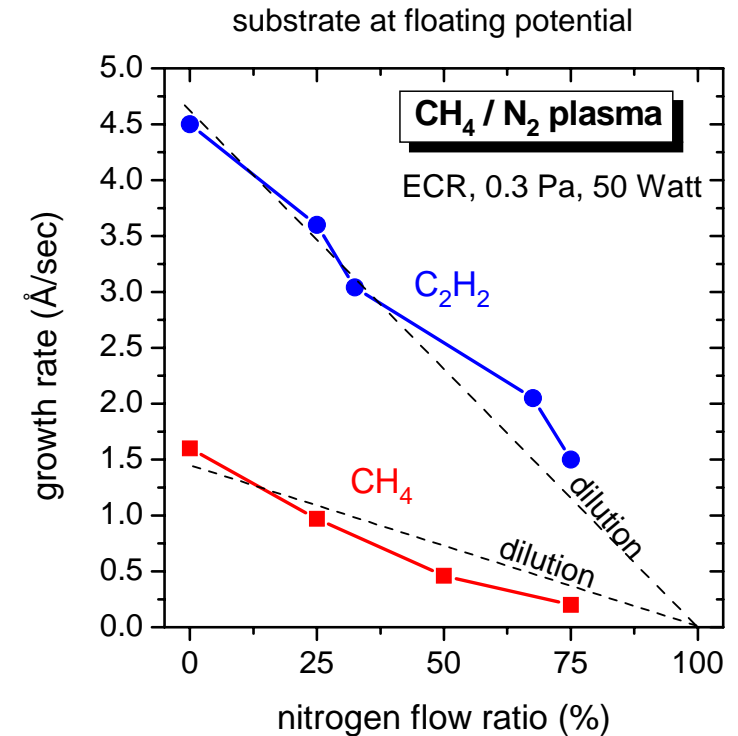
remote mass spectrometer for residual gas analysis

retarding field analyzer (RFA) at substrate position to measure the
mass integrated ion flux



A **reduction** of the deposition rate due to N_2 admixture to hydrocarbon plasmas was **confirmed** by our experiments but **does not exceed significantly the expected reduction due to dilution**.

The reduction depends on the hydrocarbon source gas, the mixing ratio, and other plasma parameters such as power, residence time etc.

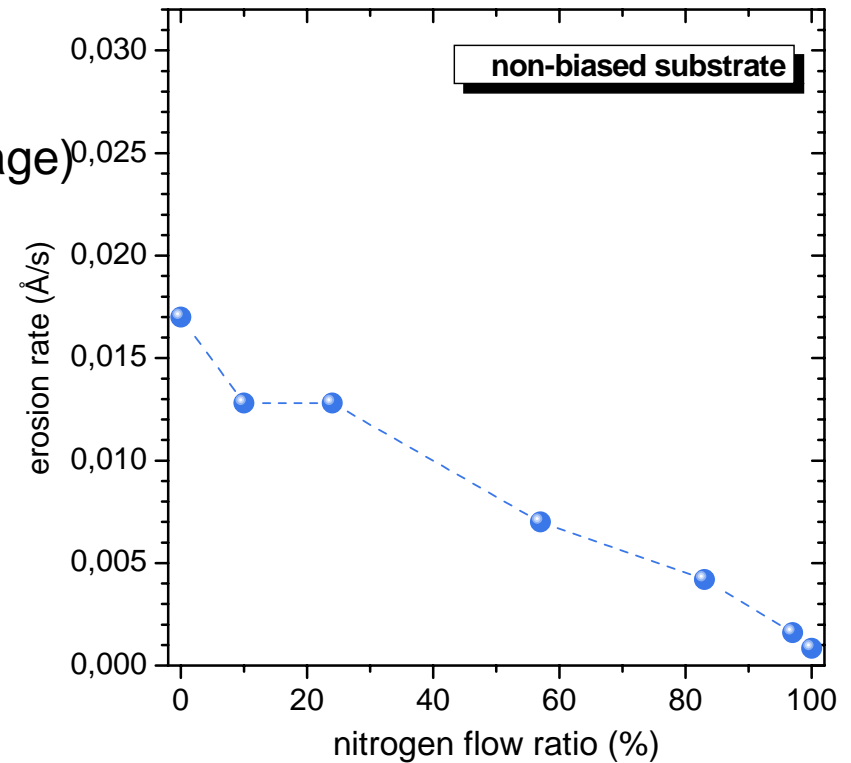


O. Durand-Drouhin et al. 2000, unpublished

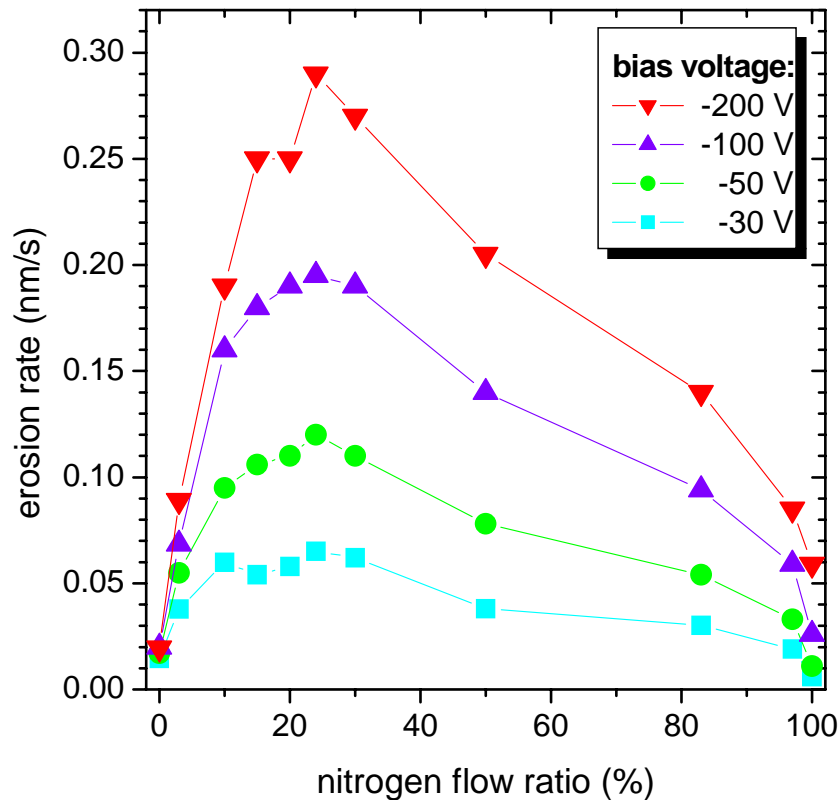
erosion of a-C:H in H₂/N₂ mixtures

Procedure:

- deposition of 100 nm hard a-C:H
 - O₂ plasma cleaning of chamber walls (cage)
 - erosion in H₂ / N₂ plasma
- decrease due to dilution ?
- nitrogen seems unreactive under these conditions



biased substrate:



- ➔ erosion rate increases with bias
- ➔ erosion rate for mixtures is higher than in pure H₂ and pure N₂ plasmas
- ➔ erosion rate increases dramatically with N₂ admixture although ion flux decreases
- ➔ clear maximum around 24 % N₂ flow ratio

➔ chemical sputtering

second step: erosion in H₂/N₂ mixtures with coated chamber walls

QMS time trace

Experimental phases:

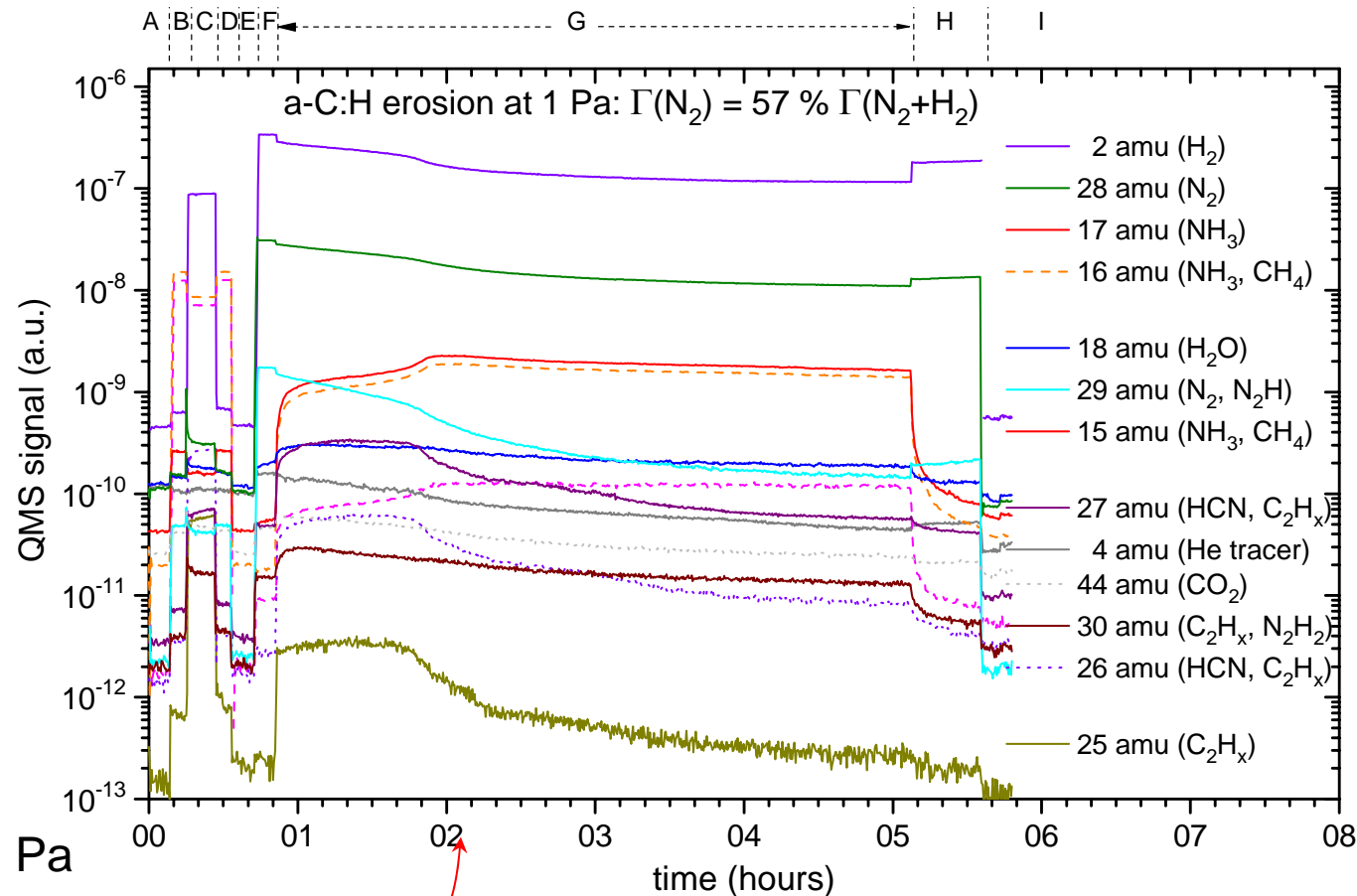
A, E, I: residual gas

B, D: CH₄ 0.2 Pa

C: CH₄ plasma

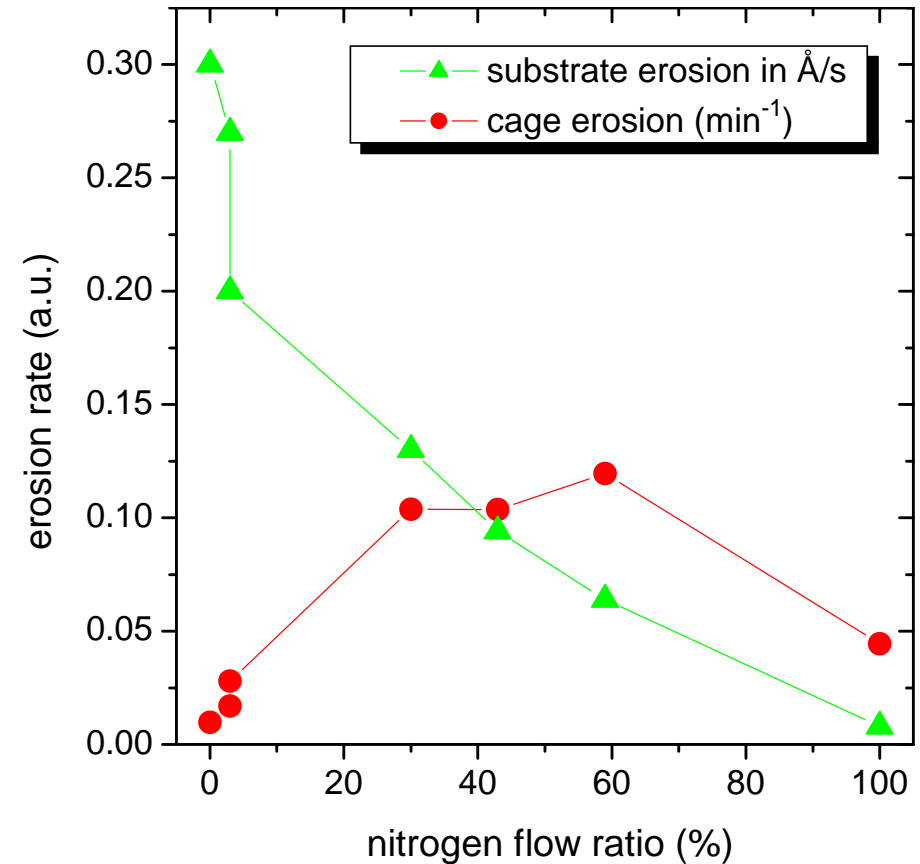
F: H₂ / N₂ mixture 1 Pa

G: H₂ / N₂ plasma



after 2 hours wall coating starts to vanish.
C_xH_y and HCN intensities drop, NH₃ intensity increases.

- ⇒ different erosion behavior in the cage and at the substrate
- ⇒ cage erosion behaves similar to biased substrate erosion (cage erosion measured as the 1/e time of the 25 amu signal)



in the same plasma different behaviour in different areas!

Procedure:

a-C:H deposition at the chamber wall
(cage)

pure N₂ plasma at 1 Pa

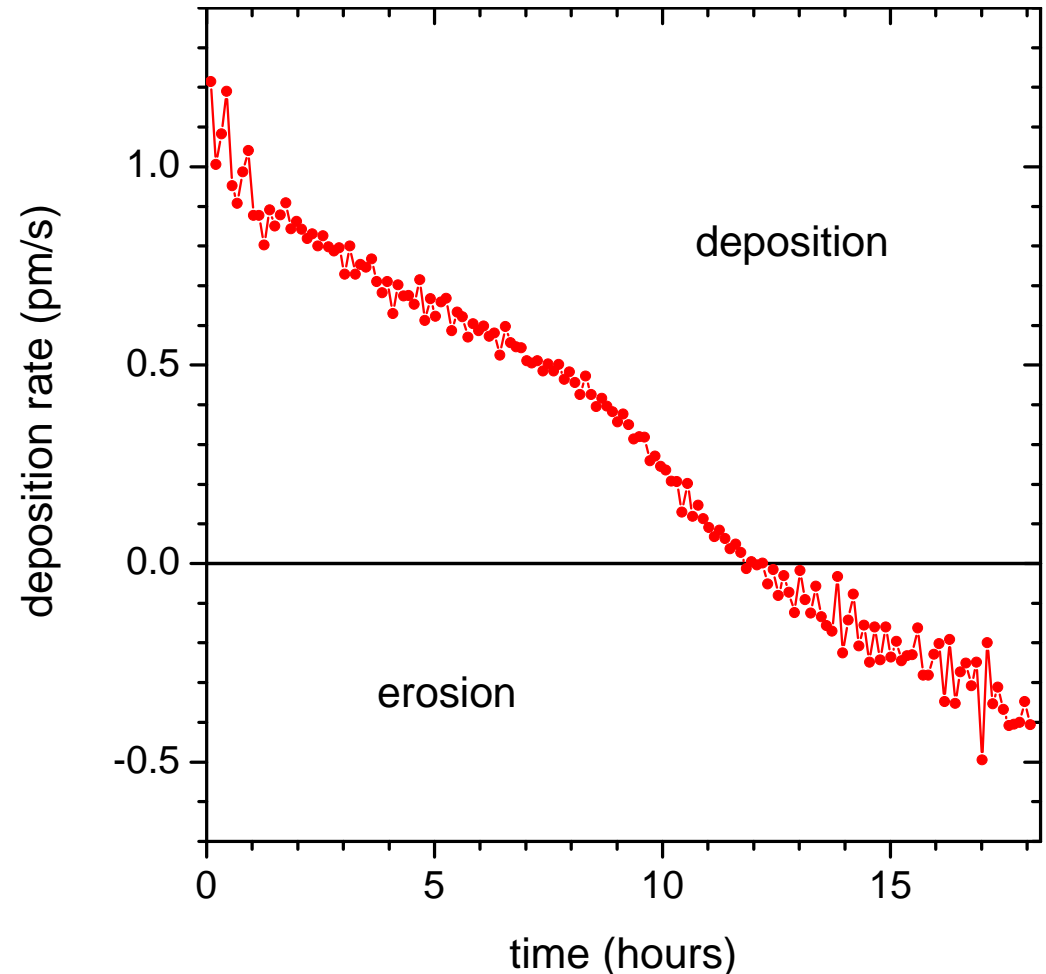
➔ redeposition of a 25 nm thick film

(XPS: $n_N / n_C = 1$; 6 % O₂)

➔ redeposition stops when cage is clean

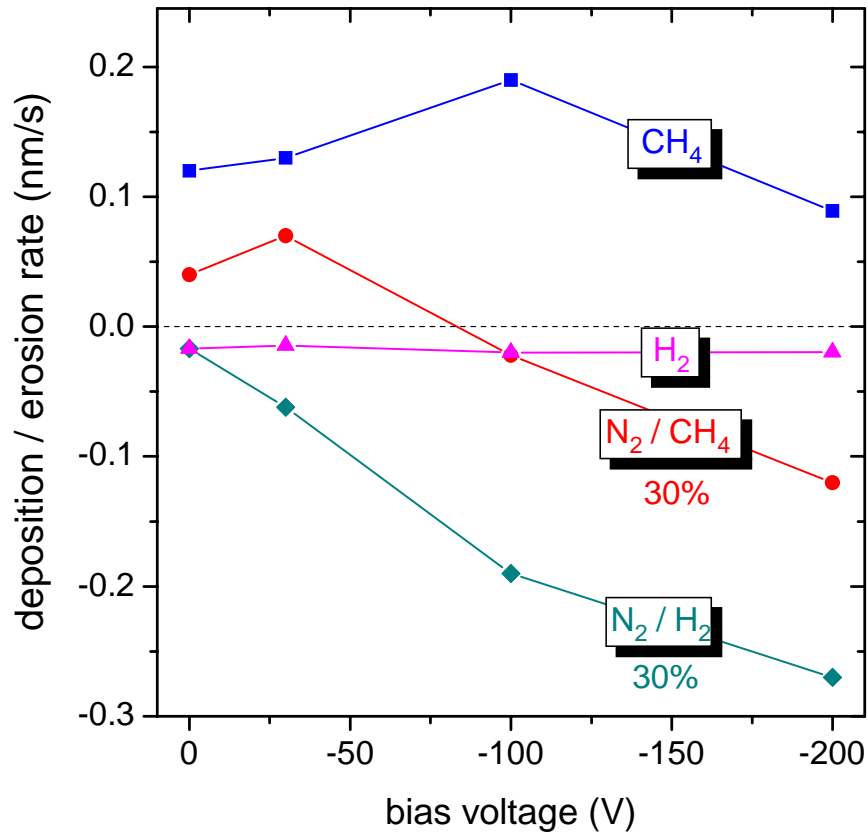
then:

➔ change over to net erosion



summary: experiments with H_2/N_2 and CH_4/N_2 mixtures

Erosion and deposition as a function of ion energy



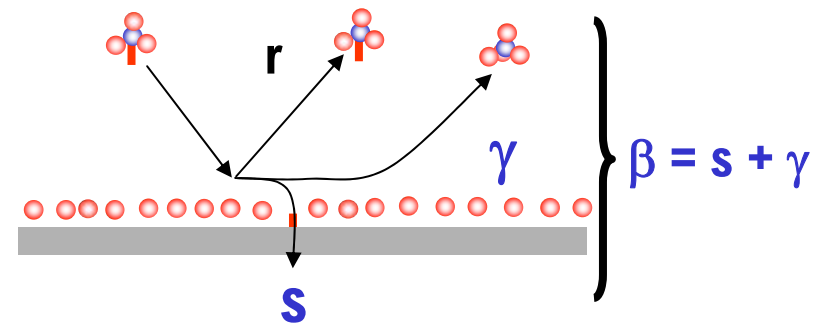
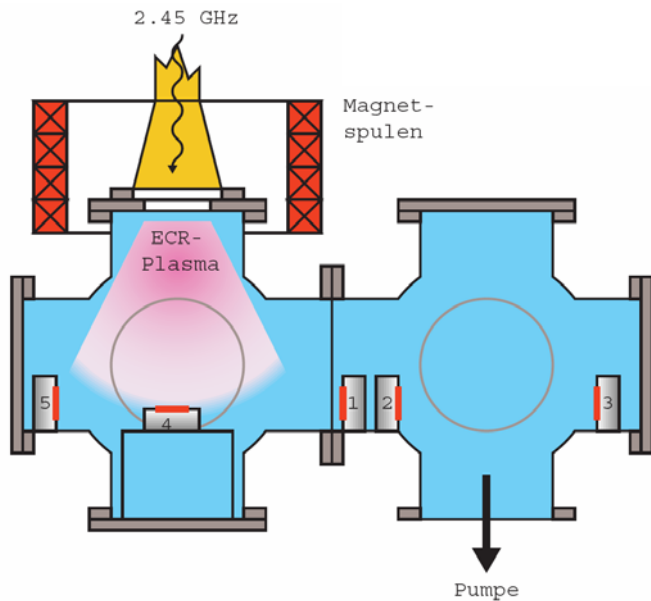
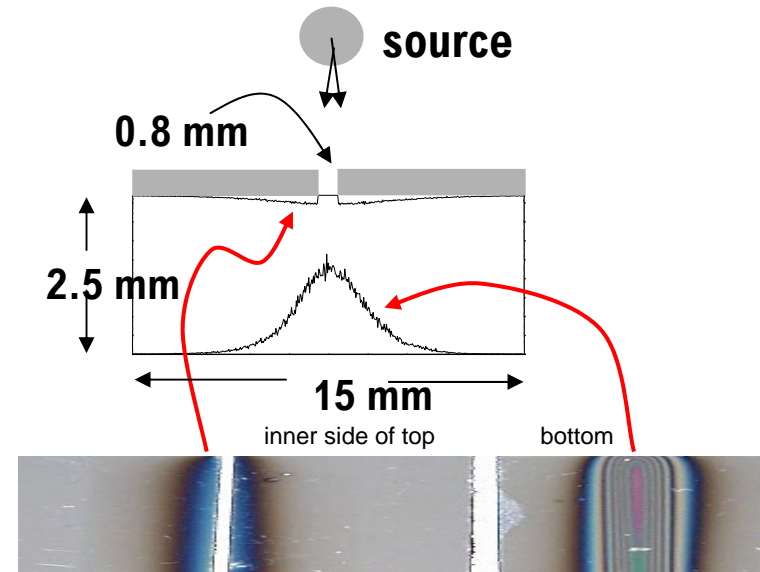
- increase of erosion rate with bias for for N₂/CH₄ and
- changeover from deposition to erosion for CH₄/N₂ plasmas at higher ion energies can **only be explained by surface effects** (the gas phase chemistry remains unchanged!)

➔ **chemical sputtering**

- **nitrogen admixture reduces carbon growth rate in low-Temp. hydrocarbon plasmas**
- **surface: nitrogen/hydrogen shows very pronounced chemical sputtering**
- **for increased ion energy growth can turn into erosion**
- **redeposition possible in pure N₂ low-T plasmas**
- **surface effects (chemical sputtering) are experimentally proven (component lifetime!)**
- **volume effects: no experimental evidence for scavenger effects**

cavity technique: surface loss probabilities of species

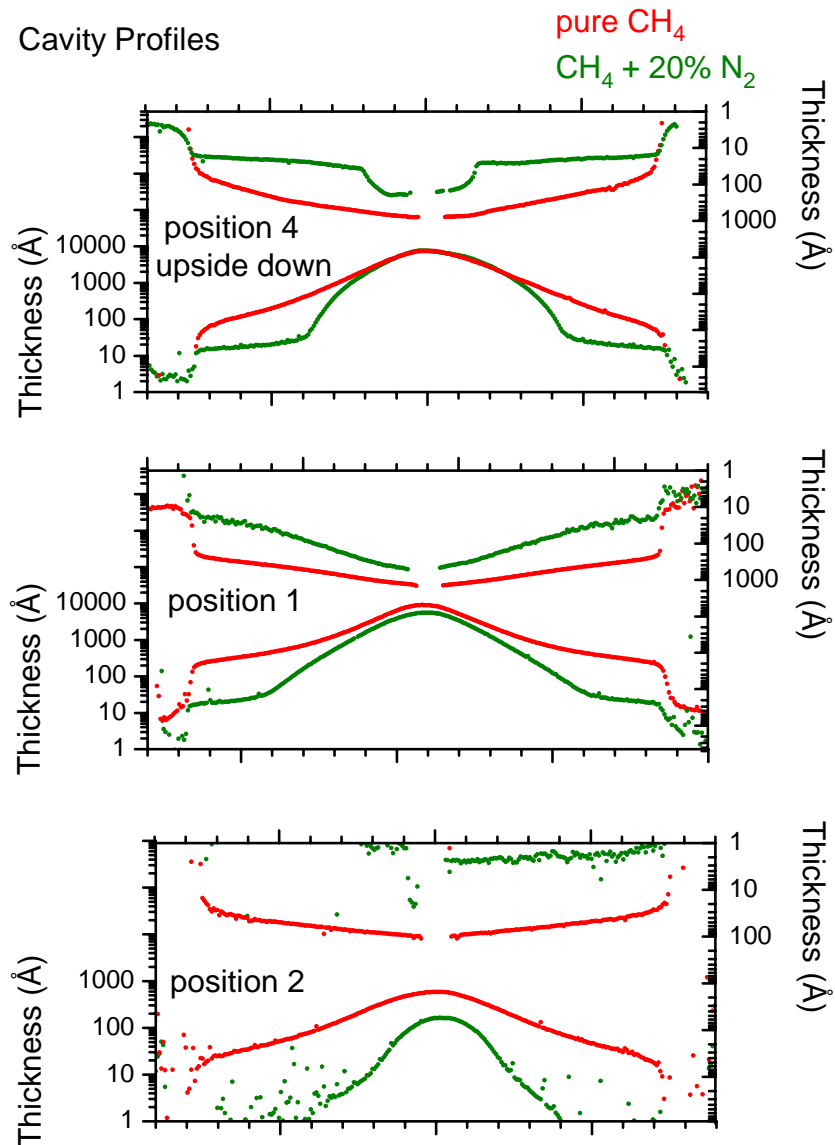
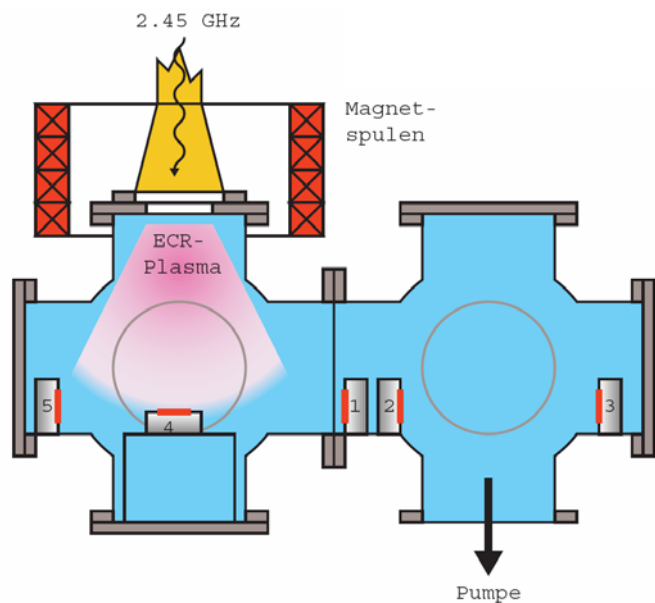
Ch. Hopf et al., J. Appl. Phys. 87 (2000)



cavity technique: surface loss probabilities of species



reduced deposition
in the wings
→ scavenging
of low β species!



Because of the increased erosion due to chemical sputtering the present data base does not allow to recommend N_2 admixture to divertor plasmas.

even if reduced film growth (e.g. by quartz microbalance) is observed in some areas of the divertor the following aspects have to be checked:

- **total carbon erosion (component lifetime)**
- **reactivity of the eroded species (redeposition)**
- **dust production**