

Problem Set 12a

Due date: January 21, 2008

Problem 45)

The four p_z orbitals of the three O atoms and of the N atom of the NO_3^- ion are oriented perpendicular to the plane of the molecule and give rise to a π system, which is occupied by two delocalized electrons. Apply the Hückel approximation for molecular orbitals and express the energies in terms of the Coulomb integrals α_O and α_N and the resonance integral β . Solve the relevant secular equations by usage of a determinant and calculate the delocalization energy of the ion. Hint: use energies which you would obtain for one hypothetical electron pair localized between N and O.

(5 points)

Problem 46)

Calculate the ratio of the Einstein coefficients of spontaneous and stimulated emission, A and B , for transitions with the following characteristics: a) 1.542 \AA (x-ray), b) 470 nm (visible) c) 2900 cm^{-1} (mid-infrared) d) 98.3 MHz (radio). Show graphically how A/B behaves as a function of frequency.

(5 points)

Problem 47)

A rotation absorption spectrum for $^1\text{H}^{35}\text{Cl}$ gas was experimentally recorded and the observed lines had the following wavenumbers: $83,32$; $104,13$; $124,73$; $145,37$; $165,89$; $186,23$; $206,60$; $226,86 \text{ cm}^{-1}$. From the latter results, calculate the moment of inertia I of the $^1\text{H}^{35}\text{Cl}$ molecule and its bond length R . Calculate the wavenumbers of the analogous lines for $^2\text{H}^{35}\text{Cl}$.

Given: reduced masses $\mu(^1\text{H}^{35}\text{Cl}) = 1,6266 \cdot 10^{-27} \text{ kg}$ and $\mu(^2\text{H}^{35}\text{Cl}) = 3,1624 \cdot 10^{-27} \text{ kg}$

(4 points)