

Structure factor: All shades of diffraction



neutron diffraction



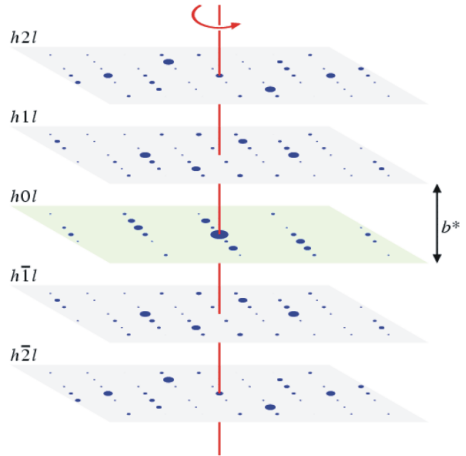
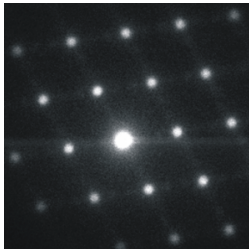
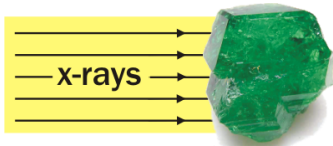
solid hydrogen



DNA team

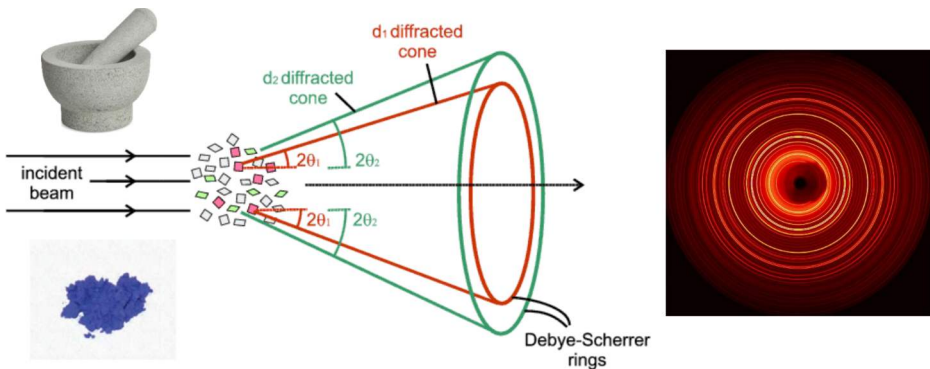


Diffraction from single crystals

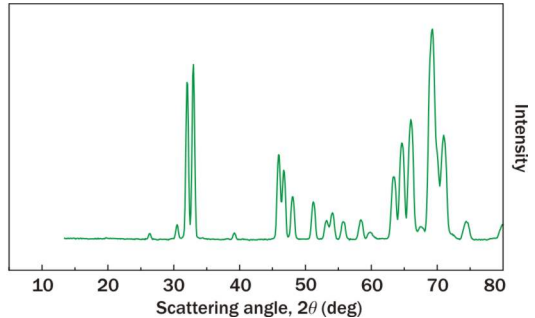
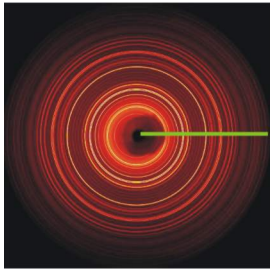


Diffraction pattern is an image of the reciprocal lattice

Diffraction from powders



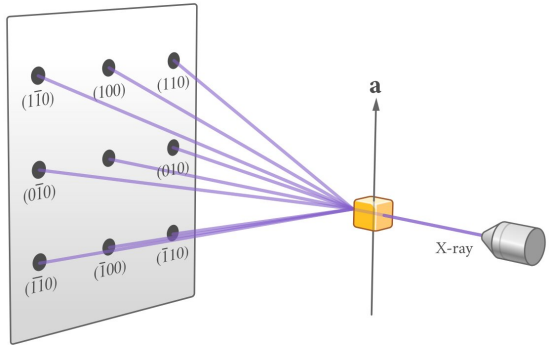
Debye-Scherrer rings, imprints of cones of scattered intensity



Debye-Scherrer rings, imprints of cones of scattered intensity

Single-crystal: **3D pattern**, powder: **1D pattern**

“Philosophy” of diffraction experiments



Positions of reflections \longrightarrow **lattice parameters**

Intensities of reflections \longrightarrow **atomic positions**

Body-centered lattice

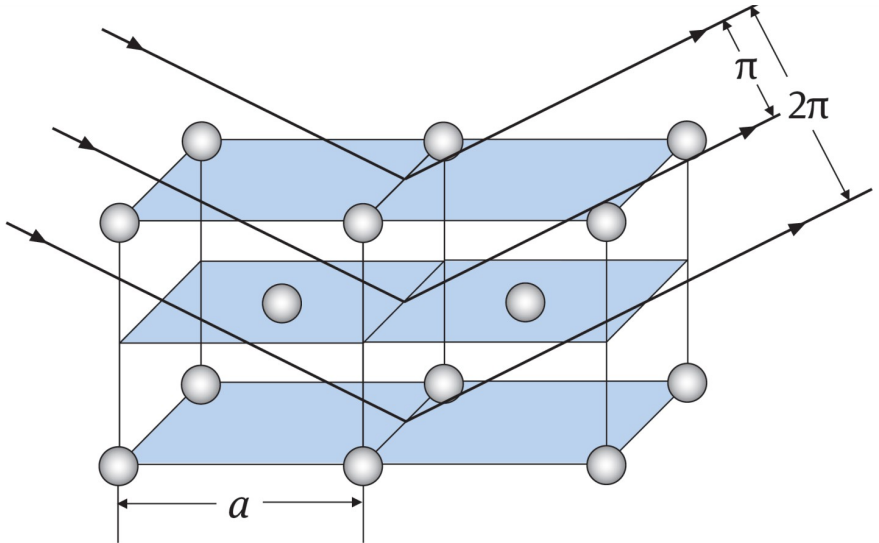
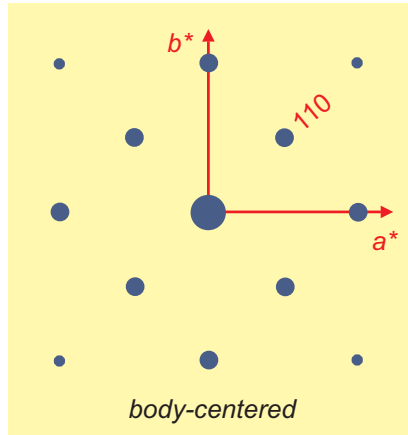
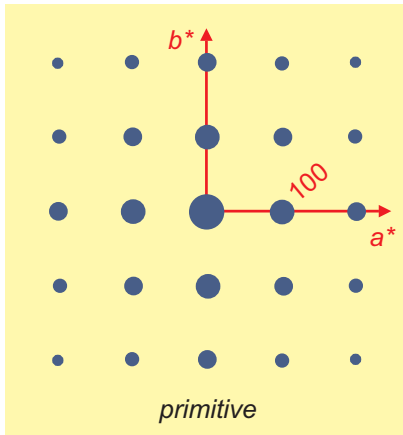


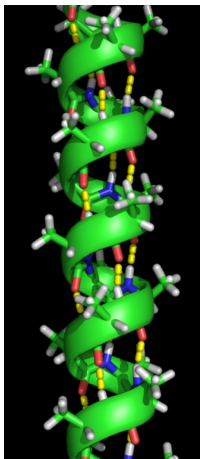
Image credit: Hunklinger, Solid-State Physics





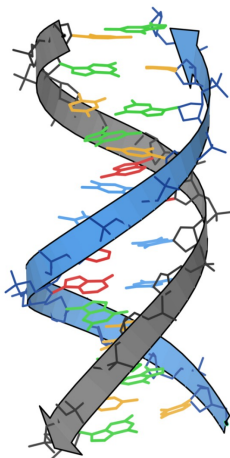
Person

DNA team



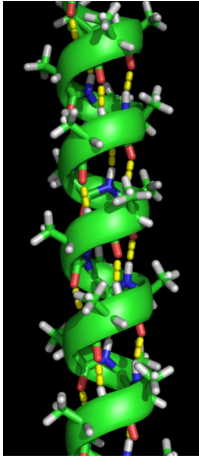
protein
 α -helix

(Pauling et al. 1951)



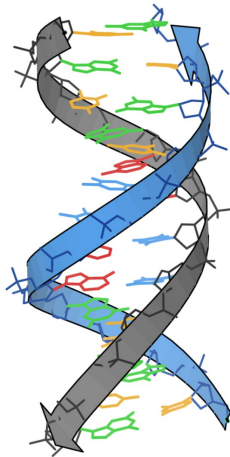
DNA
double helix

1953



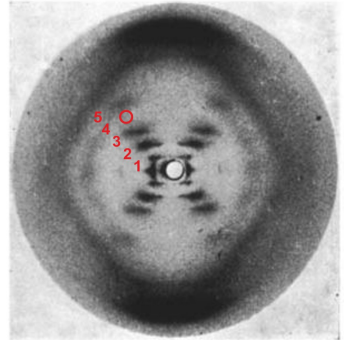
protein
 α -helix

(Pauling et al. 1951)



DNA
double helix

1953



diffraction pattern
of DNA (B-form)

Cavendish Laboratory, Cambridge



James Watson
born 1928



Francis Crick
1916–2004

King's College, London



Maurice Wilkins
1916–2004



Rosalind Franklin
1920–1958

Cavendish Laboratory, Cambridge



James Watson
born 1928



Francis Crick
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1951: first, wrong model of DNA

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1916–2004



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1920–1958

1962 Nobel prize in physiology or medicine: **Watson, Crick, Wilkins**

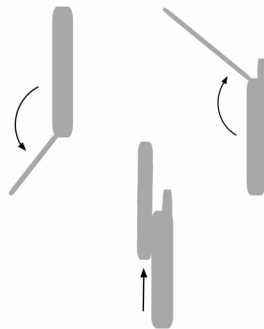


Experimental technique

neutron diffraction

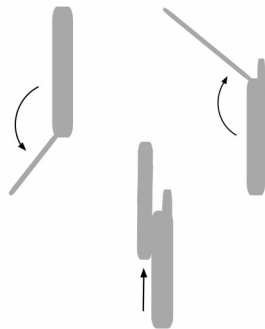
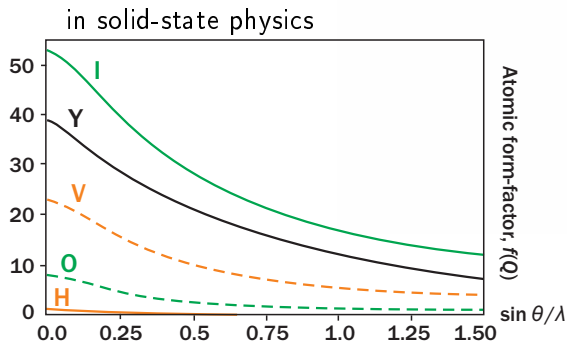
Form factor (mobile phones)

From Wikipedia, the free encyclopedia

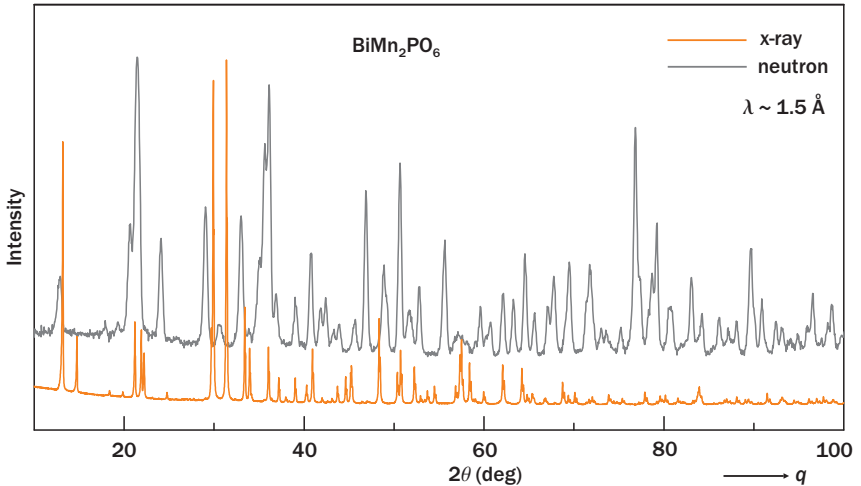


Form factor (mobile phones)

From Wikipedia, the free encyclopedia



X-ray vs. neutron form-factors



X-ray intensities decrease with q
Neutron intensities are generally q -independent

Neutron diffractometer

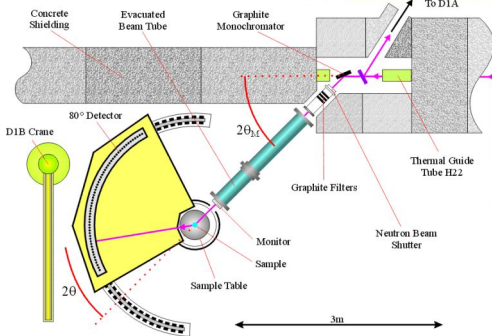
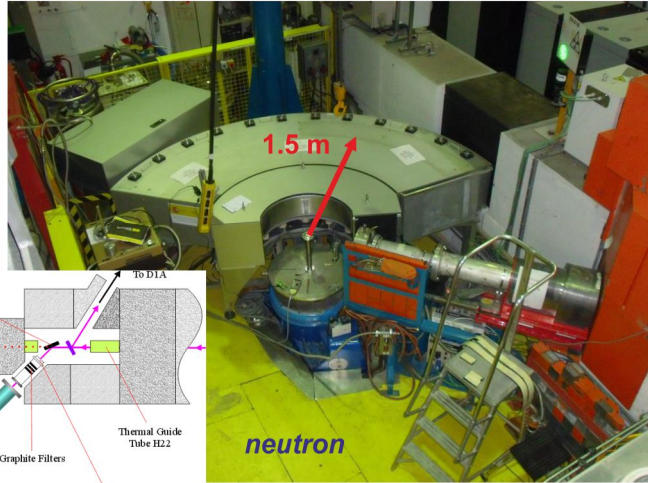
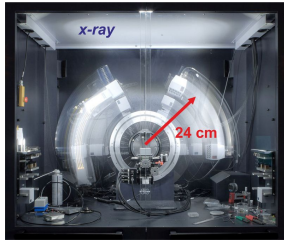


Image credits:

Kaspar Kallip (CC-BY-SA)

J. Phys. Conference Series 549, 012003 (2014)

Birkbeck College, UCL

- **Nuclear reactor:**
stable and robust neutron source,
but requires huge infrastructure
+ environmental concerns

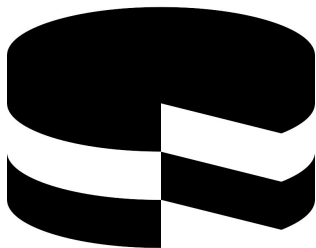
- **Spallation source:**
neutrons may arrive in pulses
less stable in general,
but more environment-friendly,
and higher flux can be achieved



Neutron sources in Europe

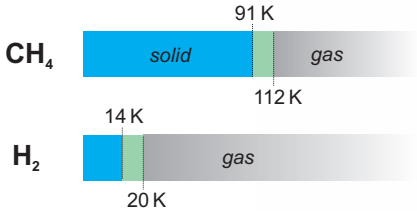


Map source: Johomaps



Material
solid hydrogen

Hydrogen storage



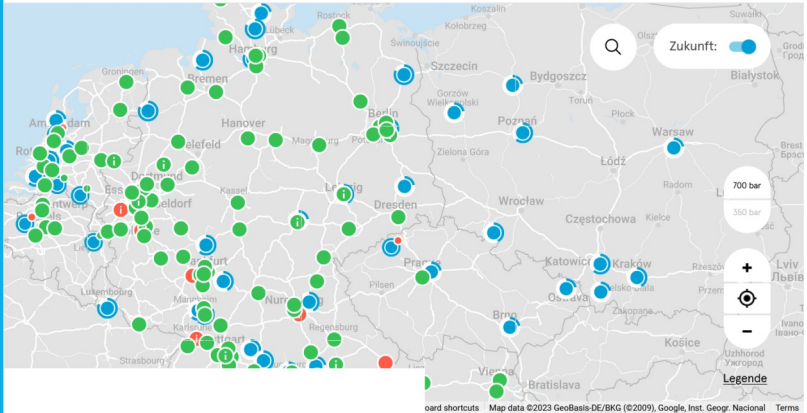
Images: Sybille Riepe and Ogiya (CC-BY-SA)

Hydrogen tank stations

H₂

Deutsch ▼ Hilfe 🔗 Anmelden

☰
Menu



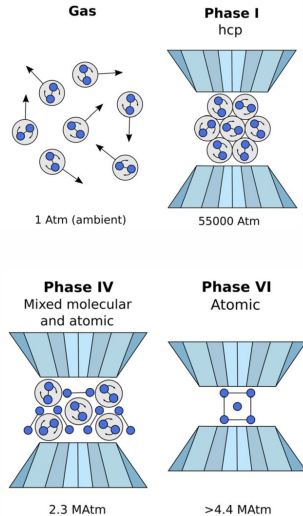
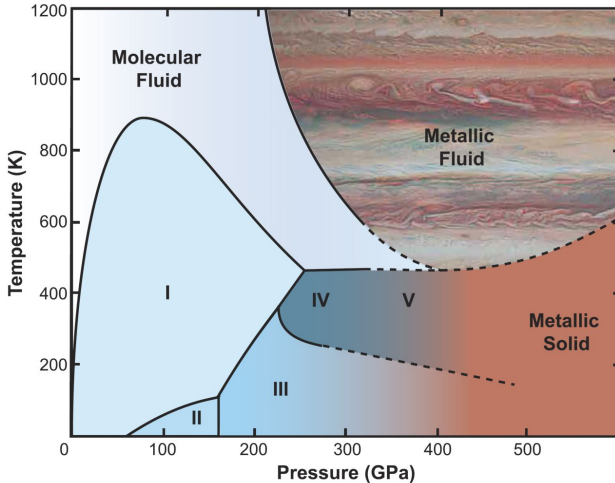
<http://h2.live>

Melting point of the elements (K)

Group	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
Period 1	H ¹ 13.99	→																H ¹	He ² 0.95
2	Li ³ 453.7	Be ⁴ 1560											B ⁵ 2349	C ⁶ 3800	N ⁷ 63.15	O ⁸ 54.36	F ⁹ 53.48	Ne ¹⁰ 24.56	
3	Na ¹¹ 370.9	Mg ¹² 923											Al ¹³ 933	Si ¹⁴ 1687	P ¹⁵ 317.3	S ¹⁶ 368.4	Cl ¹⁷ 171.6	Ar ¹⁸ 83.81	
4	K ¹⁹ 336.7	Ca ²⁰ 1115	Sc ²¹ 1814	Ti ²² 1941	V ²³ 2183	Cr ²⁴ 2180	Mn ²⁵ 1519	Fe ²⁶ 1811	Co ²⁷ 1768	Ni ²⁸ 1728	Cu ²⁹ 1358	Zn ³⁰ 692.7	Ga ³¹ 302.9	Ge ³² 1211	As ³³ 1090	Se ³⁴ 494	Br ³⁵ 265.8	Kr ³⁶ 115.8	
5	Rb ³⁷ 312.5	Sr ³⁸ 1050	Y ³⁹ 1799	Zr ⁴⁰ 2128	Nb ⁴¹ 2750	Mo ⁴² 2896	Tc ⁴³ 2430	Ru ⁴⁴ 2607	Rh ⁴⁵ 2237	Pd ⁴⁶ 1828	Ag ⁴⁷ 1235	Cd ⁴⁸ 594.2	In ⁴⁹ 429.8	Sn ⁵⁰ 505.1	Sb ⁵¹ 903.8	Te ⁵² 722.7	I ⁵³ 386.9	Xe ⁵⁴ 161.4	
6	Cs ⁵⁵ 301.7	Ba ⁵⁶ 1000	* Lu ⁷¹ 1925	Hf ⁷² 2506	Ta ⁷³ 3290	W ⁷⁴ 3695	Re ⁷⁵ 3459	Os ⁷⁶ 3306	Ir ⁷⁷ 2719	Pt ⁷⁸ 2041	Au ⁷⁹ 1337	Hg ⁸⁰ 234.3	Tl ⁸¹ 577	Pb ⁸² 600.6	Bi ⁸³ 544.7	Po ⁸⁴ 527	At ⁸⁵ 575	Rn ⁸⁶ 202	
7	Fr ⁸⁷ 300	Ra ⁸⁸ 973	** Lr ¹⁰³ 1900	Rf ¹⁰⁴ -	Db ¹⁰⁵ -	Sg ¹⁰⁶ -	Bh ¹⁰⁷ -	Hs ¹⁰⁸ -	Mt ¹⁰⁹ -	Ds ¹¹⁰ -	Rg ¹¹¹ -	Cn ¹¹² -	Nh ¹¹³ -	Fl ¹¹⁴ -	Mc ¹¹⁵ -	Lv ¹¹⁶ -	Ts ¹¹⁷ -	Og ¹¹⁸ -	
			* La ⁵⁷ 1193	Ce ⁵⁸ 1168	Pr ⁵⁹ 1208	Nd ⁶⁰ 1297	Pm ⁶¹ 1315	Sm ⁶² 1345	Eu ⁶³ 1099	Gd ⁶⁴ 1585	Tb ⁶⁵ 1629	Dy ⁶⁶ 1680	Ho ⁶⁷ 1734	Er ⁶⁸ 1802	Tm ⁶⁹ 1818	Yb ⁷⁰ 1097			
			** Ac ⁸⁹ 1323	Th ⁹⁰ 2115	Pa ⁹¹ 1841	U ⁹² 1405	Np ⁹³ 917	Pu ⁹⁴ 912.5	Am ⁹⁵ 1449	Cm ⁹⁶ 1613	Bk ⁹⁷ 1323	Cf ⁹⁸ 1173	Es ⁹⁹ 1133	Fm ¹⁰⁰ 1800	Md ¹⁰¹ 1100	No ¹⁰² 1100			

Image credit: Albris (CC-BY-SA)

Solid hydrogen

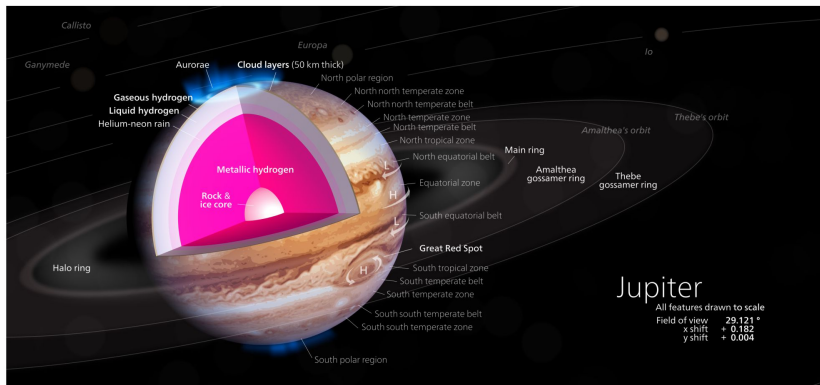


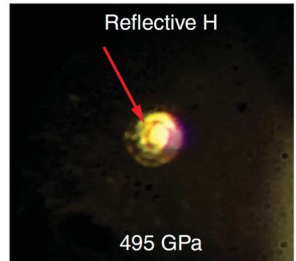
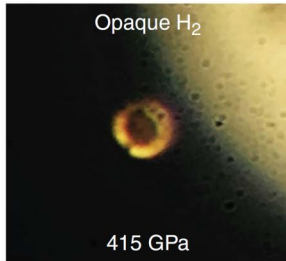
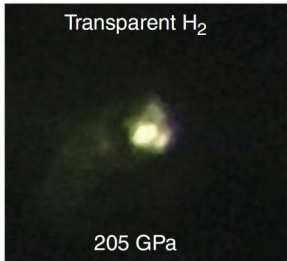
METALLIC HYDROGEN: A HIGH-TEMPERATURE SUPERCONDUCTOR?

N. W. Ashcroft

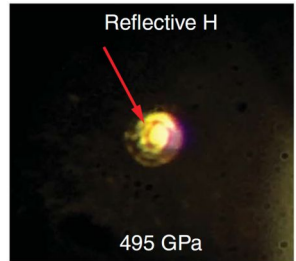
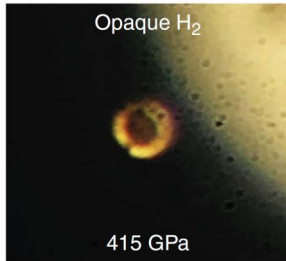
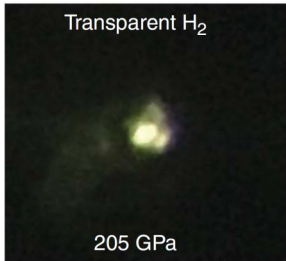
Laboratory of Atomic and Solid State Physics, Cornell University, Ithaca, New York 14850

(Received 3 May 1968)





Ranga P. Dias and Isaac F. Silvera*



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[Published: 02 February 2017](#)

Physicists doubt bold report of metallic hydrogen

[Davide Castelvichi](#)

[Nature](#) 542, 17 (2017) | [Cite this article](#)

“If they want to be convincing, they have to redo the measurement, really measuring the evolution of pressure,” says Loubeyre. “Then they have to show that, in this pressure range, the alumina is not becoming metallic.”

But Silvera says that he and Dias just wanted to get the news out before making confirmation tests, which could break their precious specimen. “We wanted to publish this breakthrough event on this sample,” he says. To preserve the material, they have kept it in the cryostat; the lab has only two cryostats, and the other is in use for other experiments. “Now that the paper has been accepted,” he says, “we’re going to do further experiments. ■

Credits: quotation from Nature 542, 17 (2017)

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World's only piece of a metal that could revolutionise technology has disappeared, scientists reveal

Ian Johnston Science Correspondent
Wednesday 22 February 2017



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The problem of metallic hydrogen awaits you!