#### All sorts of metals



#### ab initio calculations



Li-ion batteries



Marvin Cohen



Lecture 21: January 4, 2024

by Alexander Tsirlin, Leipzig University

Exp. Physics 5 - Solid State Physics, WS23/24

All sorts of metals

#### Bands vs. DOS



## Spaghetti's



## Spaghetti's



#### Van Hove singularities



#### Metal vs. insulator



R. Gross, A. Marx, Festkörperphysik

#### Semiconductors



Band gap of a semiconductor can be larger or smaller, but it's crucial that **both types of doping** should be feasible and lead to sufficiently **high mobility** of the charge



# Experimental technique *ab initio calculations*

## Density Functional Theory (DFT)



#### Experiment vs. theory



#### Experiment vs. theory



#### **Density Functional Theory**



Electron-electron interactions are replaced by an effective, exchange-correlation potential  $V_{\rm xc}$ 

**Problem:** the exact form of  $V_{xc}$  is not known, but sensible approximations exist



Bloch functions and band energies are calculated on a mesh of *k*-points within the first Brillouin zone (irreducible part)

#### Materials Project



Multiple properties are calculated for all known materials and many hypothetical ones Extensive search capabilities, prediction of thermodynamic properties, and a lot more

**Caveat:** most information is obtained from band-structure calculations and may be inaccurate or even misleading

#### Half-metals



W.E. Pickett and J.S. Moodera, Physics Today 54(5), 39 (2001)

Half-metals



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#### Half-metals



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#### Half vs. semi

#### semi-sweet



Image by Makro

 $\mathsf{Half} \neq \mathsf{Semi}$ 

Half vs. semi

#### semi-sweet



#### half-sweet



Image by Starbucks

Image by Makro

 $Half \neq Semi$ 

Half vs. semi

#### semi-sweet



#### half-sweet



Image by Starbucks

#### half a pint!



Image by KegWorks

Image by Makro

 $\mathsf{Half} \neq \mathsf{Semi}$ 

#### Semi-metals / Metalloids

Main-Group Elements s Subshell fills					Ge To					Ľ	Main-Group Elements p Subshell fills							
	1 IA						11.15	Mr.	N.									18 VIIIA
1	H	2 IIA							6				13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	He 1s <sup>2</sup>
2		4 Be 2s <sup>2</sup>		d Subshell fills							5 B 2s <sup>2</sup> 2p <sup>1</sup>	6 C 2s <sup>2</sup> 2p <sup>2</sup>	N 25 <sup>2</sup> 2p <sup>3</sup>	8 0 2s <sup>2</sup> 2p <sup>4</sup>	9 F 2s <sup>2</sup> 2p <sup>5</sup>	10 Ne 2s <sup>2</sup> 2p <sup>5</sup>		
3	11 Na 3s <sup>1</sup>	12 Mg 35 <sup>2</sup>	3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8	9 VIIIB	10	11 IB	12 IIB	13 Al 3s <sup>2</sup> 3p <sup>1</sup>	14 Si 3s <sup>2</sup> 3p <sup>2</sup>	15 P 3s <sup>2</sup> 3p <sup>3</sup>	16 S 3s <sup>2</sup> 3p <sup>4</sup>	17 Cl 3s <sup>2</sup> 3p <sup>5</sup>	18 Ar 3s <sup>2</sup> 3p <sup>8</sup>
Period A	19 K 45	20 Ca 4s <sup>2</sup>	21 Sc 3d <sup>1</sup> 4s <sup>2</sup>	22 Ti 30 <sup>2</sup> 4s <sup>2</sup>	23 V 3d <sup>9</sup> 4s <sup>2</sup>	24 Cr 3d <sup>6</sup> 4s <sup>1</sup>	25 Mn 3d <sup>5</sup> 4s <sup>2</sup>	26 Fe 3d <sup>4</sup> 4s <sup>2</sup>	27 CO 30 <sup>7</sup> 4s <sup>2</sup>	28 Ni 3d <sup>4</sup> 4s <sup>2</sup>	29 Cu 3d <sup>10</sup> 4s <sup>1</sup>	30 Zn 3d <sup>10</sup> 4s <sup>2</sup>	31 Ga 4s <sup>2</sup> 4p	32 Ge 4s <sup>2</sup> 4p <sup>2</sup>	33 As 4s <sup>2</sup> 4p <sup>3</sup>	34 Se 4s <sup>2</sup> 4p <sup>4</sup>	35 Br 4s <sup>2</sup> 4p <sup>3</sup>	36 Kr 4s <sup>2</sup> 4p <sup>8</sup>
5	37 Rb 591	38 Sr 5s <sup>2</sup>	39 Y 4d <sup>1</sup> 5s <sup>2</sup>	40 Zr 40 <sup>2</sup> 5s <sup>2</sup>	41 Nb 4d <sup>4</sup> 5s <sup>1</sup>	42 Mo 40'5s'	43 TC 4d <sup>6</sup> 5s <sup>2</sup>	44 Ru 40'5s'	45 Rh 40 <sup>6</sup> 5s <sup>1</sup>	46 Pd 40 <sup>10</sup>	47 Ag 4d <sup>10</sup> 5s <sup>1</sup>	48 Cd 4d <sup>10</sup> 5s <sup>2</sup>	49 In 5s <sup>2</sup> 5p <sup>1</sup>	50 Sn 5s <sup>2</sup> 5p <sup>2</sup>	51 Sb 5s <sup>2</sup> 5p <sup>3</sup>	52 Te 5s <sup>2</sup> 5p <sup>4</sup>	53 I 5s <sup>2</sup> 5p <sup>5</sup>	54 Xe 5s <sup>2</sup> 5p <sup>5</sup>
6	55 Cs 681	56 Ba 65 <sup>2</sup>	57 La* 5d <sup>1</sup> 6s <sup>2</sup>	72 Hf 50 <sup>2</sup> 6s <sup>2</sup>	73 Ta 5d <sup>2</sup> 6s <sup>2</sup>	74 W 5d <sup>4</sup> 6s <sup>2</sup>	75 Re 50 <sup>5</sup> 6s <sup>2</sup>	76 Os 5d <sup>4</sup> 6s <sup>2</sup>	77 Ir 5d <sup>7</sup> 6s <sup>2</sup>	78 Pt 5d <sup>9</sup> 6s <sup>1</sup>	79 Au 50 <sup>10</sup> 6s'	80 Hg 50 <sup>10</sup> 6s <sup>2</sup>	81 TI 6s <sup>2</sup> 6p <sup>1</sup>	82 Pb 65 <sup>2</sup> 60 <sup>2</sup>	83 Bi 6s <sup>2</sup> 60 <sup>3</sup>	84 Po 6s <sup>2</sup> 6p <sup>4</sup>	85 At 6s <sup>2</sup> 60 <sup>2</sup>	86 Rn 6s <sup>2</sup> 6d <sup>6</sup>
7	87 Fr 7 <i>s</i> <sup>1</sup>	88 Ra 7 <i>s</i> <sup>2</sup>	89 Ac** 6d <sup>1</sup> 7s <sup>2</sup>	104 Db 6d <sup>2</sup> 7s <sup>2</sup>	105 JI 6d <sup>a</sup> 7s <sup>2</sup>	106 Rf 6d <sup>4</sup> 7s <sup>2</sup>	107 Bh 6d <sup>5</sup> 7s <sup>2</sup>	108 Hn 60 <sup>6</sup> 7s <sup>2</sup>	109 Mt 6d <sup>7</sup> 7s <sup>2</sup>	Inr	ner-Tra f Sub	nsition shell fi	Metals	3				
*Lanthanides *Lanthanides *Lanthanides *Actinides *Actinides Th Pa U Np Pu Am Cm Bk Cf Es Fm Md No Lr									71 Lu 4/ <sup>45</sup> 50 <sup>1</sup> 65 <sup>2</sup> 103 Lr									
Metal																		
Nonmetal Image by PBworks																		

### Semi-metals (modern definition)



Electrons at the Fermi level may have very unusual properties caused by non-trivial band topology

#### Strange metals



#### Bad metals



At high T, resistivity well exceeds the Mott-loffe-Regel limit  $(I_{\rm mean} \sim a)$ 

But at low T, such "bad metals" can be still good metals (low  $\rho_{\rm dc}$ )

#### **Electronic** correlations



Each electron is dependent on other electrons in the system

#### Many-body problem

Significant repercussions for electronic properties (high-temperature superconductors, battery materials...)



# Material

Li-ion batteries

Exp. Physics 5 - Solid State Physics, WS 23/24 All sorts of metals

.

#### Li-ion batteries



#### Li-ion batteries



LiFePO<sub>4</sub> (olivine) – Nobel Prize in Chemistry 2019 but transformation is abrupt with no intermediate compositions

#### Most common correlated material

#### PHYSICAL REVIEW B 69, 201101(R) (2004)

#### Phase separation in Li<sub>x</sub>FePO<sub>4</sub> induced by correlation effects

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We report on a <u>significant failure</u> of the local density approximation (LDA) and the generalized gradient approximation (GGA) to <u>reproduce the phase stability and thermodynamics</u> of mixed-valence Li<sub>x</sub>FePO<sub>4</sub> compounds. Experimentally, Li<sub>x</sub>FePO<sub>4</sub> compositions ( $0 \le x \le 1$ ) are known to be unstable and phase separate into LiFePO<sub>4</sub> and FePO<sub>4</sub>. However, first-principles calculations with LDA/GGA yield energetically favorable intermediate compounds and hence no phase separation. This qualitative failure of LDA/GGA seems to have its

> Phase separation in batteries is driven by electronic correlations Straight-forward DFT calculations do not work

Information from Materials Project

LiFePO <sub>4</sub>		
mp-19017	Energy Above Hull	0.000 eV/atom
	Space Group	Pnma
	Band Gap	3.92 eV
	Predicted Formation Energy	-2.477 eV/atom
	Magnetic Ordering	Ferromagnetic
	Total Magnetization	4.00 µB/f.u.

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## Person

#### Marvin Cohen



Marvin Cohen born 1935

#### 1953-1957: bachelor in Berkley, not accepted for master program

- 1958-1963: master and PhD in Chicago theory work based on electronic structure calculations for real materials
- 1963-1964: very short postdoc at Bell Labs
- since 1964: professor at Berkley
- 1960-70's: optical properties of semiconductors
- since 1970's: high-pressure transformations of solids
- 1980's: superconductors
- since 1990's: carbon allotropes, nanotubes



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#### Cohen vs. Kohn



photo by Bill Brooks (Physics Today)

Marvin Cohen pseudopotentials pioneer of electronic-structure and DFT methods in solids

#### Walter Kohn

density-functional theory Nobel Prize in Chemistry 1998

#### **Pseudopotentials**



#### Superconducting semiconductor



SrTiO<sub>3</sub> is a mundane wide-gap semiconductor

#### Superconducting semiconductor



 $SrTiO_3$  is a mundane wide-gap semiconductor but  $SrTiO_{3-\delta}$  should be metallic and may become superconducting

#### Superconducting semiconductor

